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Solutions Manual for Mathematics for Physical Chemistry

vii This book provides solutions to nearly of the exercises and problems in Mathematics for Physical Chemistry, fourth edition, by Robert G Mortimer This edition is a revision of a third edition published by Elsevier/Academic Press in

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CHAPTER 13 CHEMICAL KINETICS

CHAPTER 13: CHEMICAL KINETICS 343 From the first set of data: $320 \times 10^{-1} \text{ M/s} = k(150 \text{ M})$ $k = 0.213 \text{ s}^{-1}$ What would be the value of k if you had used the second or third set of data? Should k be constant? 1318 Strategy: We are given a set of concentrations and rate data and asked to determine the order of the reaction and the initial rate for specific concentrations of X and Y

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